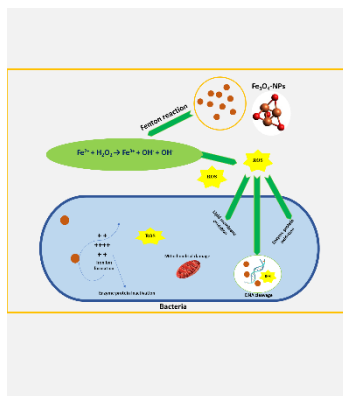


Antibacterial Activity of Fe₃O₄-NPs/H₂O₂ Combination: A Promising Approach for Efficient Bacterial

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The study explored the catalytic role of ferromagnetic nanoparticles (Fe₃O₄-NPs) in generating reactive oxygen species (ROS) with H₂O₂. It aimed to assess biocidal efficacy against four bacteria strains. Fe₃O₄-NPs synthesis involved a straightforward method ensuring uniform size and morphology. Optimal ROS synthesis conditions were determined through a 4-factor Box-Behnken design, yielding pH 6, 42 °C temperature, and specific Fe₃O₄-NPs/H₂O₂ concentrations (160 µg/mL/0.136 µg/mL). The Fe₃O₄-NPs/H₂O₂ combination induced ROS production, damaging bacterial cells via oxidative stress, showcasing potential for pathogen control and combating antibiotic resistance.

Introduction

The emergence of antibiotic-resistant bacteria poses a global health threat, necessitating innovative strategies to combat infections [1]. The WHO's research agenda on antimicrobial resistance (AMR) outlines critical priorities to address by 2030, aligning with CDC reports on antibiotic-resistant infections causing significant mortality [2]. Traditional antibiotics' declining efficacy underscores the urgency for novel antibacterial approaches. Iron oxide magnetic nanoparticles (Fe₃O₄-NPs) have emerged as promising antibacterial agents due to their unique properties and catalytic activity [3]. Their application in the Fenton-like reaction, generating reactive oxygen species (ROS) in the presence of H₂O₂, demonstrates potent antibacterial effects by inducing oxidative stress in bacteria [4]. This study investigates Fe₃O₄-NPs/H₂O₂ antibacterial activity, focusing on structural characteristics and ROS-mediated bacterial inactivation mechanisms.

Material and Methods

The study employed a rotatable central composite design (RCCD) to investigate Fe₃O₄-NPs' catalytic ROS production with H₂O₂ and their biocidal potential, analyzing factors like Fe₃O₄-NPs and H₂O₂ concentrations. A subsequent Box-Behnken design introduced additional factors (temperature, pH) to optimize ROS synthesis, evaluating Fe₃O₄-NPs and H₂O₂ concentrations, with tests conducted in microplates using chromogenic substrate TMB to identify the Fenton reaction in the presence of ROS. Absorbance measurements at specific wavelengths and stability checks were performed, employing response surface

methodology for data analysis.

Results and Discussion

Figure 1 depicts the response surfaces generated through response surface methodology for these designs. In Figure 5A, the response surface for one bacterium tested in the initial experiment is displayed, aiming to investigate the occurrence of the Fenton reaction without prior optimization of variables like pH and temperature. The absence of interaction between Fe₃O₄-NPs and H₂O₂ indicates that the biocidal effect was solely due to H₂O₂, suggesting no Fenton reaction took place, leading to a lack of ROS generation with potential biocidal effects. The experimental data aligned well with a quadratic equation, where Y represents inhibition (1) or growth (0) of the microorganism tested, showcasing the predominant influence of factor A (H₂O₂) supported by statistical values like R², Std. Dev., and C.V.

In the second experimental design, the focus shifted to determining optimal operating conditions for pH, temperature, Fe₃O₄-NPs concentration, and H₂O₂ concentration in the Fenton reaction. Figure 2 displays the UV-vis spectrum from preliminary tests confirming ROS production through reagent reactions. Characteristic peaks at 450 nm (cationic radical presence) and 370/650 nm (charge transfer complex) indicated successful reactions. Absorbance readings at 370 nm were most intense and stable, leading to a quadratic model fitting where Y represents absorbance values. Statistical parameters like R², C.V., and Std. Dev. were significant, supporting the model's validity. Notably, quadratic parameters had no impact on the process.

The response surface in Figure 1B post fitting experimental data at 370 nm under specific pH and

temperature conditions highlighted a significant interaction between Fe₃O₄-NPs and H₂O₂, favoring high nanoparticle concentrations and low H₂O₂ concentrations for optimal conditions. Additionally, Figure 1C illustrated a perturbation diagram showcasing the influence of all factors in ROS synthesis under constant pH and temperature conditions, emphasizing the importance of Fe₃O₄-NPs concentration followed by pH and temperature in influencing the process.

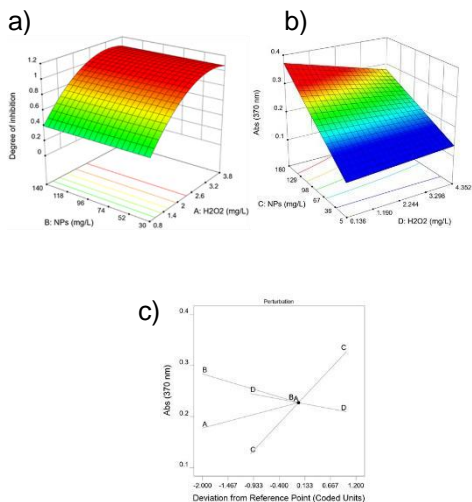


Figure 1. a) Response surface obtained from the adjustment of the experimental data of a Rotatable Central Composite Design with 2 factors in which the biocidal effect of the interaction between H₂O₂ (A) and Fe₃O₄-NPs (B) is evaluated for the case of *P. fluorescens*. b) Response surface obtained from the experimental data of a 4-factor Box-Behnken Design in which the capacity of H₂O₂ (D) and Fe₃O₄-NPs (C) to synthesize reactive oxygen species (ROS) is evaluated. The diagram shows the absorbance at the wavelength (370 nm) at which the charge transfer complexes characteristic of ROS are detected for experimental conditions of 42 °C and pH 6. c) Perturbation diagram in which the influence of the different factors is represented by setting the values of factors A and B at T: 42 °C and pH 6 respectively and where C and D represent the concentrations in mg/L of Fe₃O₄-NPs and H₂O₂.

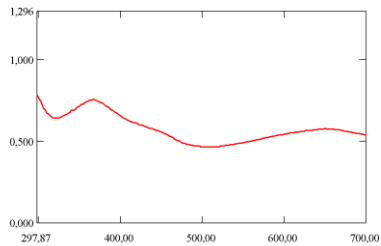


Figure 2. UV-vis spectrum where the three characteristic peaks of the reactive oxygen species obtained in the Fenton reaction produced between Fe₃O₄-NPs and H₂O₂ are identified when reacting with TMB during the preliminary tests at the following conditions: 0.00025 M Fe₃O₄, 0.1 M H₂O₂ and 0.00208 M TMB (in pH 4.5 buffer).

Conclusions

In this work, the proposed experimental design demonstrated the effectiveness of Fe₃O₄-NPs in promoting the Fenton reaction and the production of ROS with biocidal capacity. This practical application allowed for the reduction of the H₂O₂ concentration while achieving the same inhibitory effect against the four microbial strains studied. Furthermore, this work introduced the novelty of establishing the optimal operating conditions required to achieve the highest production of ROS, and consequently the highest biocidal response in combinations between Fe₃O₄-NPs and H₂O₂. These conditions were determined to be: temperature 42 °C, pH 6, and concentrations of 0.136 µg/mL for H₂O₂ and 160 µg/mL of Fe₃O₄-NPs.

Acknowledgments

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